Chemistry

Alphabet

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From chemical elements to complex chemical systems, from inorganic to organic to physical chemistry the study of chemistry involves various laboratorial techniques. Some of these are, electron microscope, magnetic resonance and spectroscopy.
This book aimed at covering among other topics, analysis techniques, chemical bonds, energy storage, inorganic chemistry (C), inorganic matters, medicine and drugs, organic C, physical C and programmable drugs.

Alphabet.

h																	ḧ
l	b̈											b	c	n	o	f	n̈
n̂	m											a	s̋	p	s	c̃	ạ
k	ĉ	ṡ	t	v	c̣	m̃	f̈	ċ	n̋	c̋	z	g	g̈	à	s̈	ḅ	k̇
r	s̈	y	z̈	ǹ	ṁ	ṫ	f̋	ȓ	ṕ	â	ć	ĩ	s̃	s̀	ẗ	i	x
c̀	b̂	j	h̄	t̂	w	r̈	ò	ị	p̄	ă	ĥ	t̃	p̀	b̋	ṗ	ā	r̃
f̣	r̂	q	r̄	d	ŝ	b̀	h̀	m̄									
		c̈	ṗ	ń	p̌	š	e	ǵ	b̄	d̂	ḣ	ẹ	ť	ỳ	ĺ		
		t̀	p̂	u	n̄	p̋	ă	č	b̆	c̄	è	f̌	ḿ	ṅ	ḷ		

Example. [a] (aluminium atom); [ĉ] (calcium atom); [h] (hydrogen atom); [ḧ] (helium atom); [f̈] (iron atom); [hc̃] (hydrochloric acid); [h_2o] (water); [h_2o_2] (hydrogen peroxide); [hs] (hydrogen sulphide); [m] (magnesium atom); [n̂] (sodium atom); [n̂c̃] (table salt); [o_2] (oxygen molecule); [s̋] (silicon atom); [t] (titanium atom)

Bibliography

T. W. Graham Solomons, *Organic Chemistry* (1976), 6^{th} ed., 1996
Gareth Thomas, *Medicinal Chemistry*, 2^{nd} ed., 2007

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